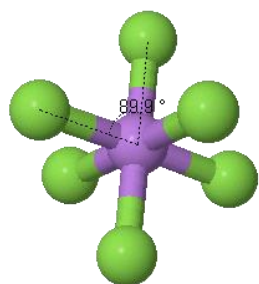
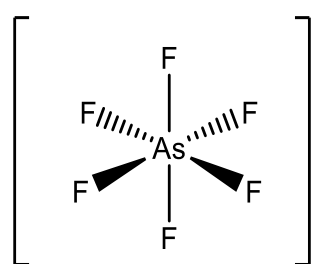


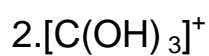
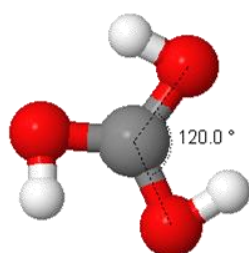
15- Identify the Shapes of These Molecules Answers



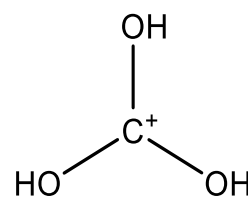
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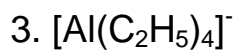
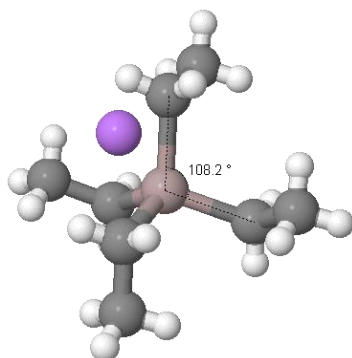
Octahedral



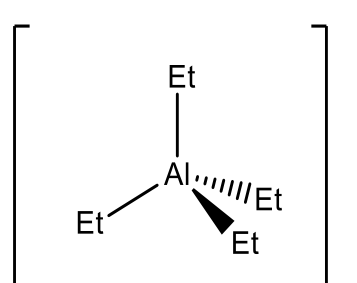
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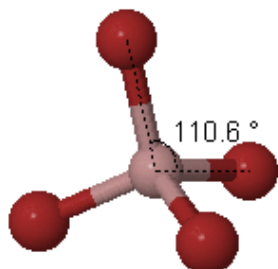
Trigonal Planar



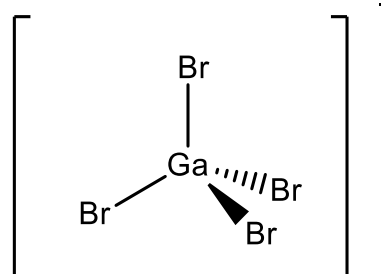
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Tetrahedral

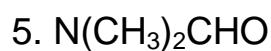
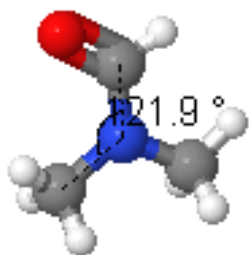


Recode:
FIGKOV

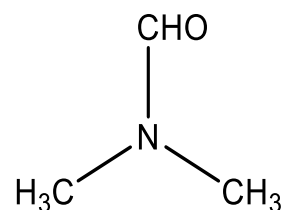


Tetrahedral

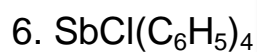
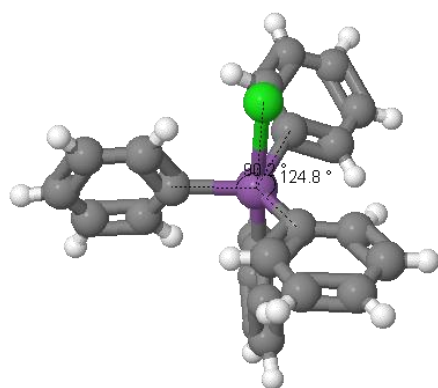
15- Identify the Shapes of These Molecules Answers



Refcode:
ABEQEF

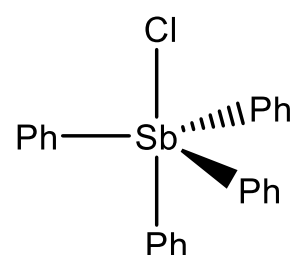


Trigonal Planar



Refcode:
BEJFIF

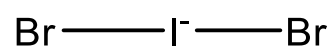
The Cl is in an axial position as it is smaller than Ph and larger groups prefer the equatorial position



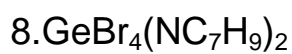
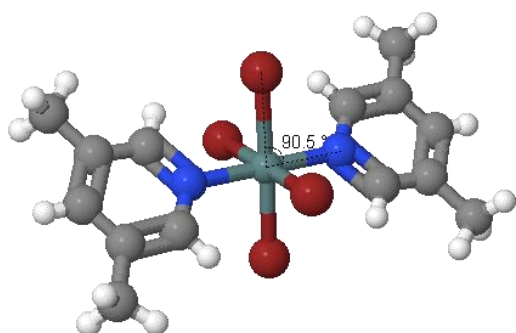
Trigonal Bipyramidal



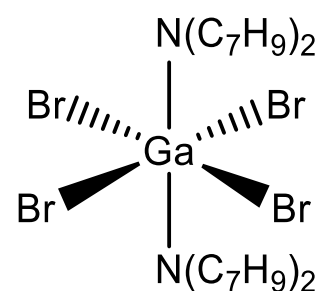
Refcode:
QIRHUU



Linear



Refcode:
BOMQOJ



Octahedral